

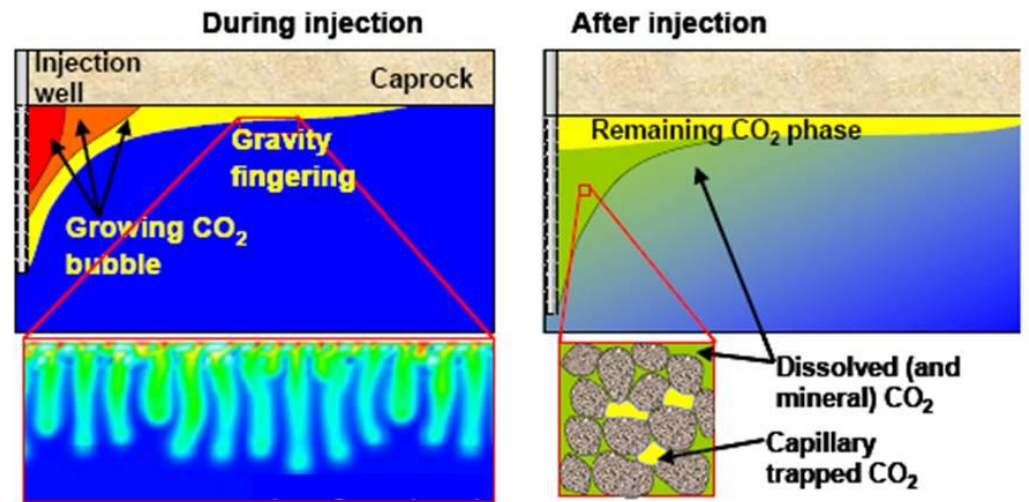
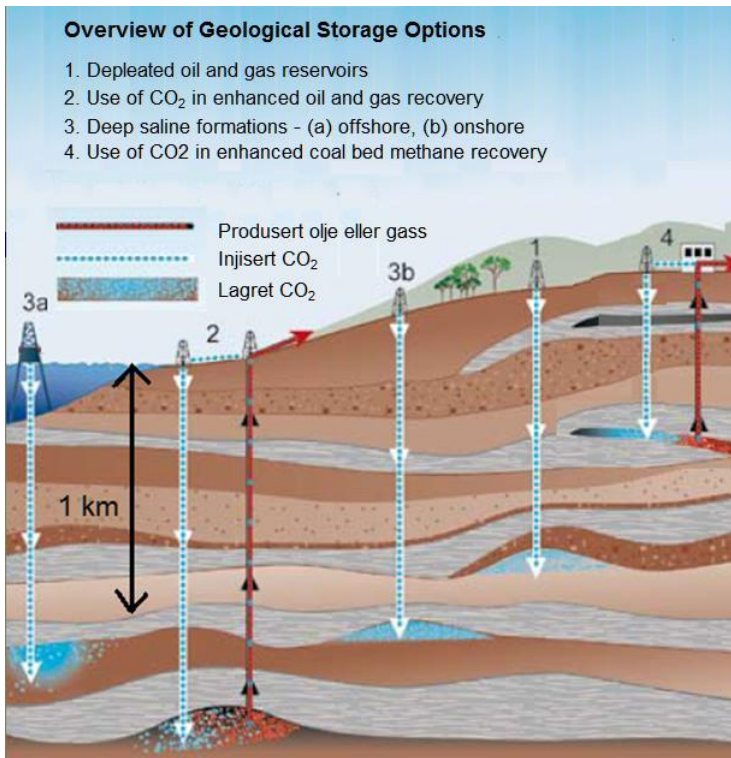
**AMPHOS**<sup>21</sup>  
SCIENTIFIC AND STRATEGIC ENVIRONMENTAL CONSULTING

# Reactive transport and convective mixing during CO<sub>2</sub> migration in a saline aquifer

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**Orlando Silva**

# The problem

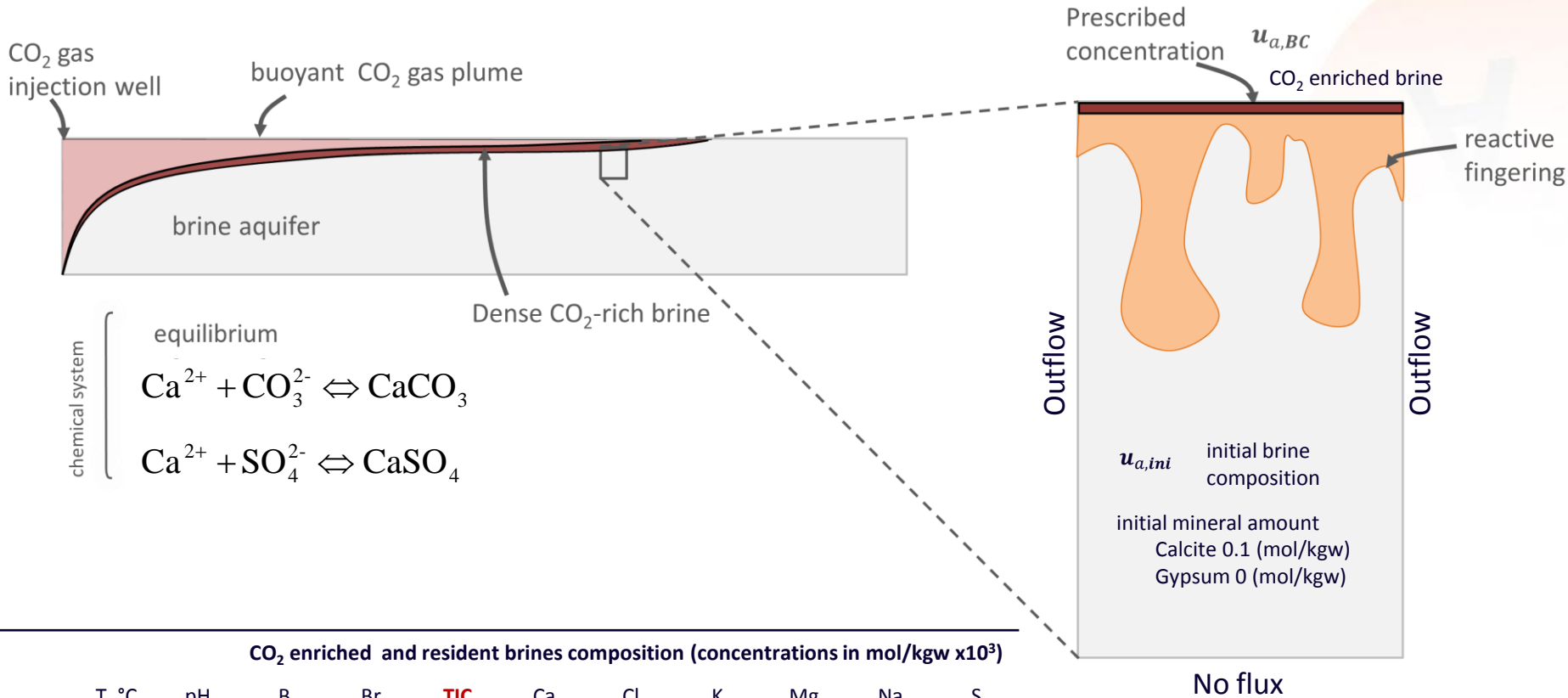
The conventional concept of CO<sub>2</sub> geological storage consists of injecting supercritical CO<sub>2</sub> into a saline aquifer at a sufficient depth so that the CO<sub>2</sub> has a high density (about 800 kg/m<sup>3</sup>) and therefore occupies a reduced volume. This requires pressures higher than 80 bars, which explains that injection must be carried out at depths greater than 800 m.



During injection CO<sub>2</sub> floats and extends through the top of the aquifer below the caprock. CO<sub>2</sub> dissolves into the resident brine increasing its density. This causes the brine to sink by means of fingering process. The resultant acid mixture reacts with the rock, leading to CO<sub>2</sub> precipitation as carbonate.

# Objective and conceptual model

Study the interaction between convective mixing and reactive transport processes related to CO<sub>2</sub> dissolution into the resident brine of a carbonated aquifer.



CO<sub>2</sub> enriched and resident brines composition (concentrations in mol/kgw x10<sup>3</sup>)

	T, °C	pH	B	Br	TIC	Ca	Cl	K	Mg	Na	S
$u_{a,BC}$	25	3.261	1.465	1.036	1443	53.51	593.6	14.34	26.26	470.7	24.31
$u_{a,ini}$	25	6.980	1.465	1.036	1.370	53.51	593.6	14.34	26.26	470.7	24.31

# Methodology

## 1. Governing equations

### Flow in porous media

$$\frac{\partial(\phi\rho_l)}{\partial t} = \nabla \cdot \rho_l \mathbf{q}_l$$

The coupled system of equations is solved using the widely spread Sequential Non Iterative Approach (SNIA), which is based on the Operator Splitting concept.

### Reactive transport

$$U_a \frac{\partial(\phi\rho_l c_a)}{\partial t} + U_d \frac{\partial(\phi\rho_l c_d)}{\partial t} + U_m \frac{\partial((1-\phi)\rho_m c_m)}{\partial t} = U_a L_l(c_a) + US_k^t r_m(c)$$

### Density dependent on temperature and composition

$$\rho_l = \rho_l(c_a, T)$$

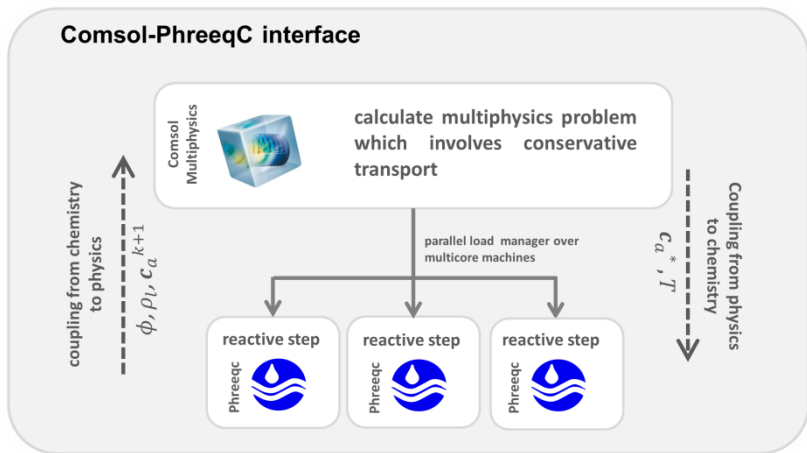
### Porosity and permeability updated from mineral precipitation

$$\phi^{k+1} = \phi^k + \omega_l^w \sum_{m=1}^{N_m} V_m c_m^{k+1}$$

$$\mathbf{\kappa} = \mathbf{\kappa}_0 \left( \frac{\phi}{\phi_0} \right)^3$$

## 2. Numerical tool

The interface Comsol-Phreeqc (iCP) was used to perform the simulations. iCP combines the key capabilities of Phreeqc and Comsol in a single reactive transport simulator.



# Interface Comsol-Phreeqc (iCP)

## Comsol Multiphysics v4.3



Code that allows simulating different physical phenomena based on differential equations using finite element methods.

## Phreeqc v2

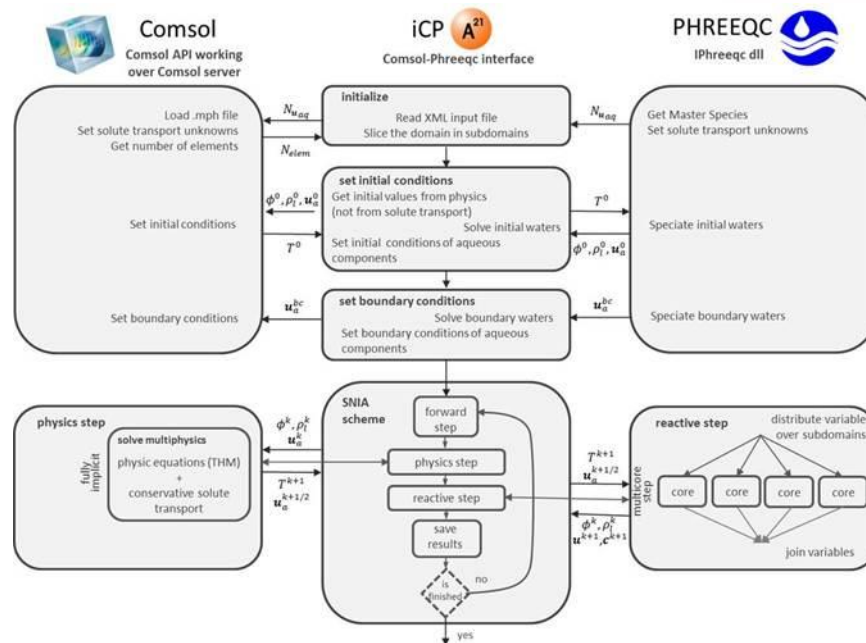


Code to simulate **chemical reactions** and transport processes in aqueous systems.

## iCP



Interface developed at Amphos 21 that couple Phreeqc to Comsol, thus allowing to simulate a wide range of **reactive transport** problems in hydrogeochemistry, hydrology, mining, etc.





# Used Comsol characteristics

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The model was implemented using the Darcy's Law (Subsurface Flow module, Fluid Flow), and Solute Transport (Chemical Species Transport module) physics.

## Darcy's Law

## Solute transport

The screenshot shows the 'Darcy's Law' physics interface in COMSOL. The left pane shows the 'Model Builder' tree with 'Darcy's Law (dl)' selected under 'Physics'. The main pane shows the 'Equation' section set to 'Time dependent' with the following equations:

$$\frac{\partial}{\partial t}(\rho \epsilon_p) + \nabla \cdot (\rho \mathbf{u}) = Q_m$$
$$\mathbf{u} = -\frac{K}{\mu}(\nabla p + \rho g \nabla D)$$

The 'Gravity Effects' section is set to 'On'. The 'Elevation' is defined as 'y-d' (m) and the 'Acceleration of gravity' is 'g\_const' (m/s<sup>2</sup>). The 'Dependent Variables' section shows 'Pressure' is 'p'.

The screenshot shows the 'Solute Transport' physics interface in COMSOL. The left pane shows the 'Model Builder' tree with 'Solute Transport (esst)' selected under 'Physics'. The main pane shows the 'Equation' section set to 'Study controlled' with the following equation:

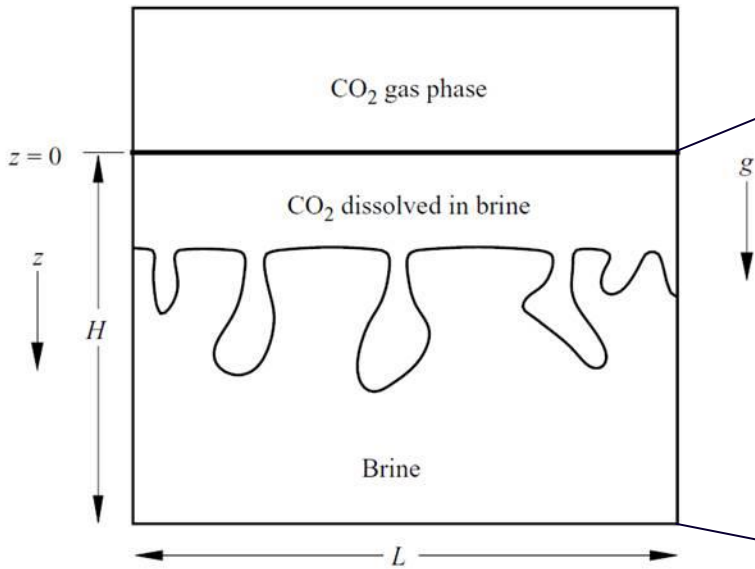
$$P_{1,j} \frac{\partial c_i}{\partial t} + P_{2,j} + \nabla \cdot \mathbf{\Gamma}_i + \mathbf{u} \cdot \nabla c_i = R_i + S_i$$

The 'Dependent Variables' section shows 'Number of species' is 12 and 'Concentrations' are 'B\_out', 'Br\_out', 'C\_out', and 'Ca\_out'.

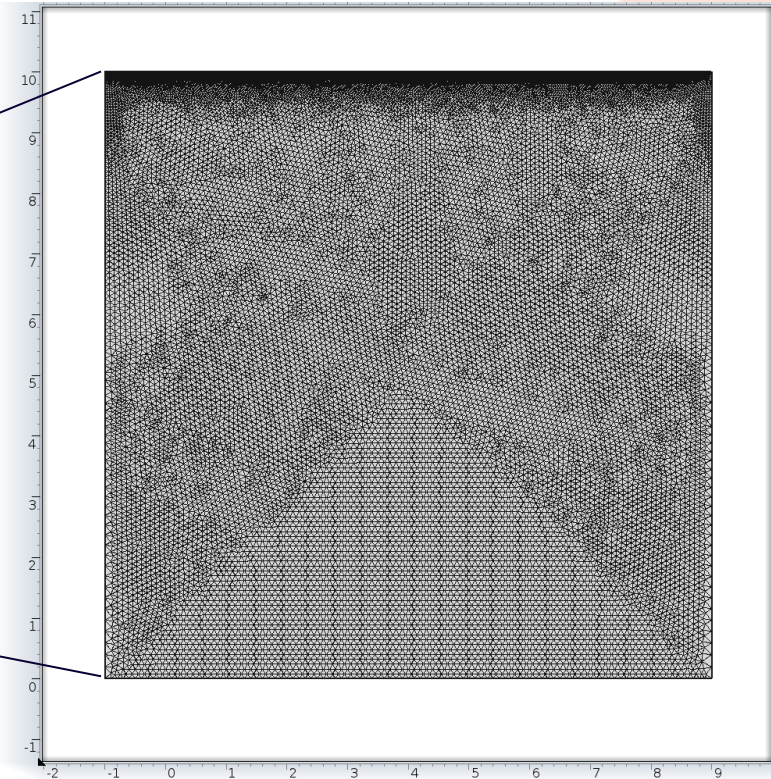
# Onset of convection during CO<sub>2</sub> injection: Single-phase flow with reactive transport. *Fingering effects*

## Conceptualization

Riaz et al., 2006. *J. Fluid Mech.*,  
vol 548, pp. 87-111



## Comsol simulation

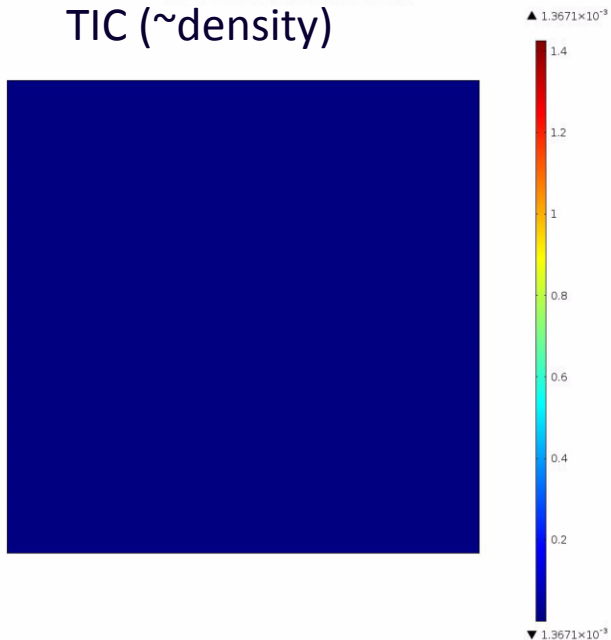


41572 triangular elements  
Simulation time 9.5 years ( $3 \times 10^8$  s)

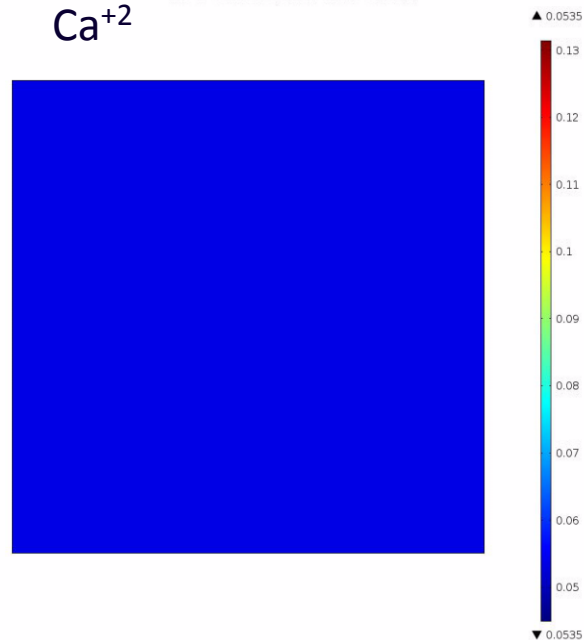
$t_{ini}, s$	$t_{end}, s$	Nº time steps
0	$3 \times 10^6$	1
$3 \times 10^6$	$10^7$	9
$10^7$	$10^8$	99
$10^8$	$3 \times 10^8$	200

# Fingering effects: impact on mineral dissolution

Time=1 Surface: Dependent variable C\_out (1)  
TIC (~density)

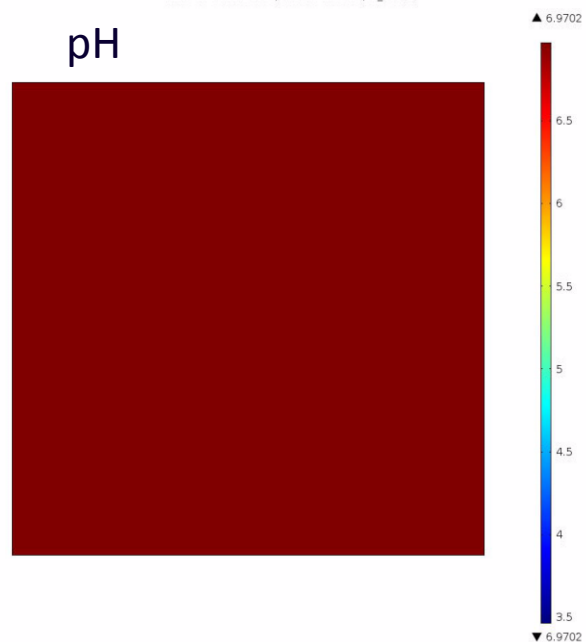


Time=1 Surface: Dependent variable Ca\_out (1)  
Ca<sup>2+</sup>



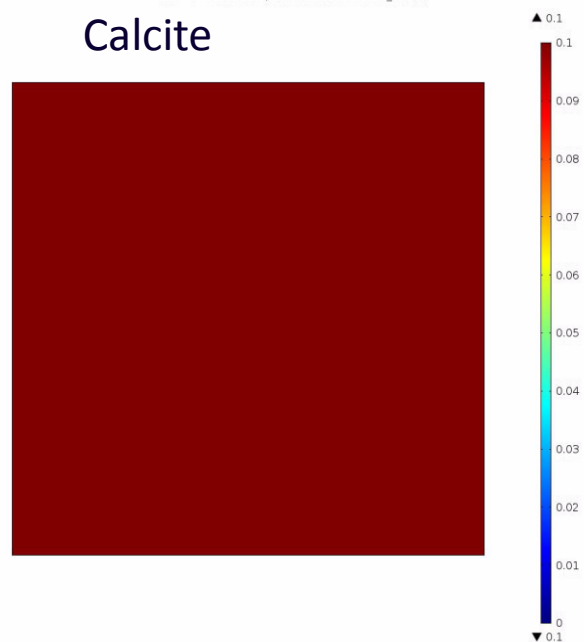
Time=1 Surface: Dependent variable pH\_out (1)

pH



Time=1 Surface: Dependent variable Calcite\_out (1)

Calcite

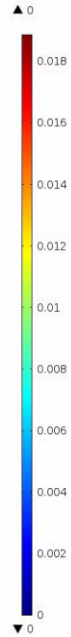
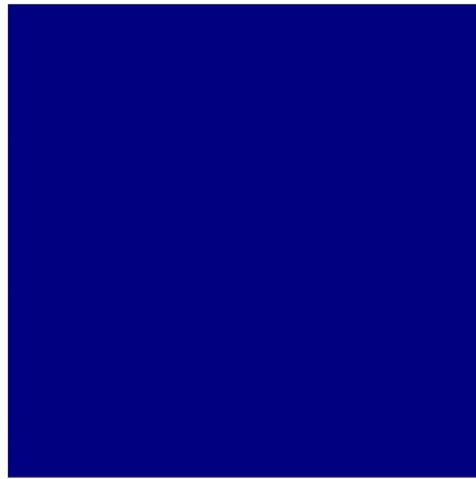




# Mineral precipitation/dissolution and porosity changes

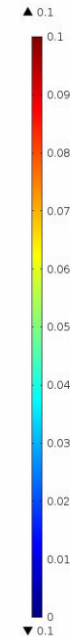
Time=1 Surface: Dependent variable Gypsum\_out (1)

## Gypsum



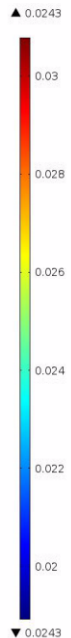
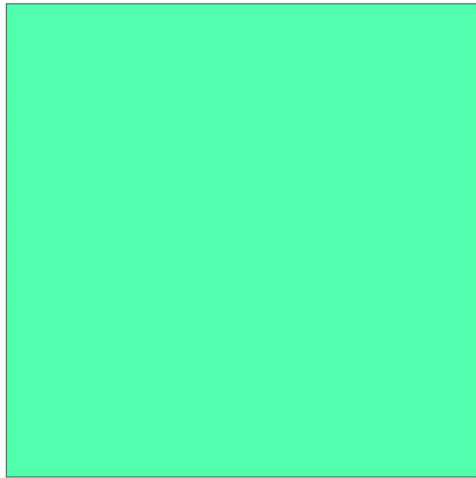
Time=1 Surface: Dependent variable Calcite\_out (1)

## Calcite



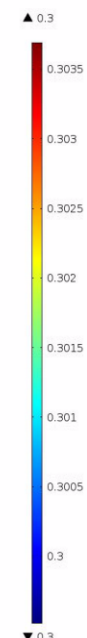
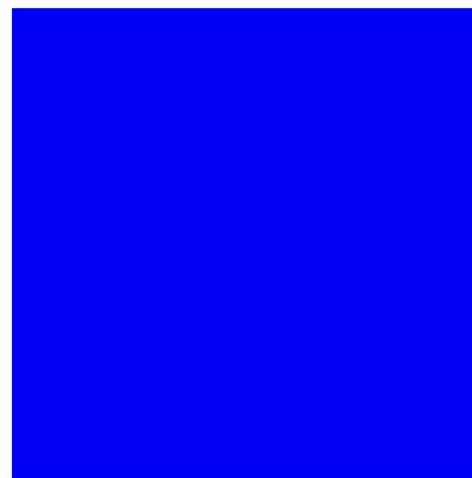
Time=1 Surface: Dependent variable S\_out (1)

## S



Time=1 Surface: Dependent variable porosity\_out (1)

## Porosity



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# Conclusions

- ❑ Physical instability results in fingering of brine rich in CO<sub>2</sub>.
- ❑ Fingering of acidic CO<sub>2</sub>-rich brine can lead to:
  - Heterogeneous distributions of Ca<sup>2+</sup> and SO<sub>4</sub><sup>2-</sup>
  - Non-uniform calcite dissolution and gypsum precipitation patterns.
- ❑ Flow and reactive transport are strongly coupled: porosity and permeability are significantly affected by dissolution/precipitation patterns; fingering enhances mixing and promotes chemical reactions.
- ❑ The interface iCP (Comsol-Phreeqc) is a powerful tool to evaluate mineralization and dissolution changes that occur during CO<sub>2</sub> injection and storage.

# Acknowledgements



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