Impact of Electrode Surface/Volume Ratio on Li-ion Battery Performance

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Introduction



[1] Long, 2004



[2] Gowda, 2011



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Simulation Structure



-ve electrode \rightarrow Carbon +ve electrode \rightarrow LiMn2O4 Electrolyte \rightarrow 1:2 EC:DMC (with 1.0 M LiPF6 salt)



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Simulation Results:



Normalized Li+ ion concentration during discharge process for different thickness of electrodes



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Normalized Li+ concentration for different pairs of electrodes during discharge process at various voltage levels.



Discharge and Energy Capacity





Normalized energy capacity and discharge time for different number of electrode pairs.

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Thermal Model



Battery structures used for thermal modeling with pairs of current collectors

$$R^{+} = \rho^{+} \frac{L}{WT} \qquad R^{-} = \rho^{-} \frac{L}{WT}$$
$$P = I^{2}R^{-} + I^{2}R^{+} = I^{2}R$$

 ρ^+ Resistivity of +ve current collector ρ^- Resistivity of -ve current collector L, W, T Length width and Thickness of current collector

With n pairs of electrodes thickness = T / nCurrent through each electrode i = I / n

Power consumption by each current collector is

$$p = (I/n)^2 nR = P/n$$



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Power Dissipation and Temperature Distribution



Temperature distribution of negative current collectors in battery pairs of electrodes with the same total current.



Normalized power consumption of each current collector, and normalized temperature versus the number of current collectors.



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With the Increase in number of electrode pairs increases energy capacity but it saturates after a certain number



With the Increase in number of current collector power consumption by each of them decreases as well as the temperature of it.



Reference

[1] Long, J. W.; Dunn, B.; Rolison, D. R.; White, H. S., "Three-dimensional battery architectures". *Chemical Reviews* **104** (10), pp. 4463-4492 (2004).

[2] S. R. Gowda, et. al., "Building Energy Storage Device on a Single Nanowire," *Nano Lett.*, **11**, pp. 3329–3333, (2011)

[3] B. Scrosati, and J. Garche, "Lithium batteries: Status, prospects and future," *Journal of Power Sources*, **195**, pp. 2419–2430, (2010)

[4]. P. Taheri, et. al., "Electrical Constriction Resistance in Current Collectorsof Large-Scale Lithium-Ion Batteries," *Journal of The Electrochemical Society*, **160** (10) A1731-A1740 (2013).

[5] Arico, A. S.; Bruce, P.; Scrosati, B.; Tarascon, J. M.; Van Schalkwijk, W., Nanostructured materials for advanced energy conversion and storage devices. *Nature Materials* **4** (5), pp. 366-377 (2005).

[6] D. R. Rolison, et, al., "Multifunctional 3D nanoarchitectures for energy storage and conversion," *Chem. Soc. Rev.*, **38**, pp. 226–252, (2009)

[7] B. Scrosati, "Challenge of portable power," *Nature*, **373**, pp. 557-558 (1995)

[8] Winter, M., and Brodd, R.J.: 'What are batteries, fuel cells, and supercapacitors?', Chemical Reviews, **104** (10), pp. 4245-4269, (2004).

[9]. G.F. Kennell, R. W. Evitts, "Two- Dimensional Lithium-Ion Battery Modeling with Electrolyte and Cathode Extensions," *Advances in Chemical Engineering and Science*, **2** (4), pp. 423-434, (2012).

[10] N. Ding, J. Xu, Y. X. Yao, G. Wegner, X. Fang, C. H. Chen, *et al.*, "Determination of the diffusion coefficient of lithium ions in nano-Si," *Solid State Ionics*, **180**, pp. 222-225 (2009).

[11] S. I. Lee, U. H. Jung, Y. S. Kim, M. H. Kim, D. J. Ahn, and H. S. Chun, "A study of electrochemical kinetics of lithium ion in organic electrolytes," *Korean Journal of Chemical Engineering*, **19**, pp. 638-644 (2002).



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