



COMSOL Multiphysics Simulations of Graphene Chemical Vapor Deposition (CVD) Growth

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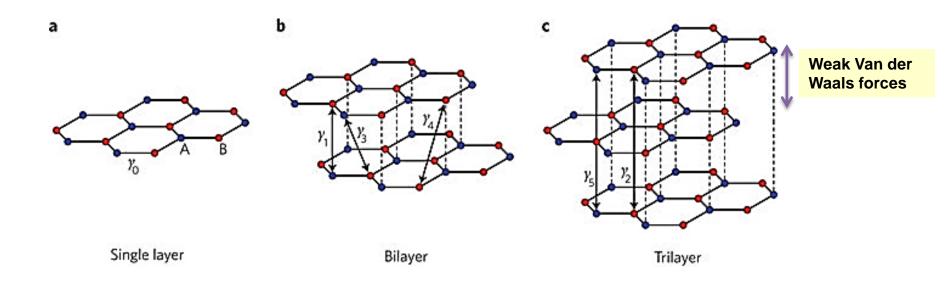
Outline

- Graphene and Graphene properties
- COMSOL Multiphysics Simulations of Graphene CVD Growth Mechanism Using Cu
- COMSOL Computational Methods
- Simulated Results and Discussion
- Conclusion



What is graphene?

Graphene is a single atomic layer, first isolated in 2004, organized in a two dimensional hexagonal (honeycomb) lattice structure



Graphene (sp2 bonded carbon atoms)

The carbon-carbon bond length in graphene is approximately 0.142 nm.

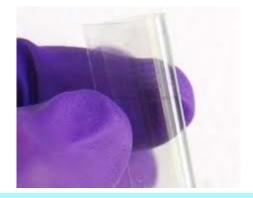
Graphite

Graphite itself consists of many graphene sheets stacked together



Graphene Properties

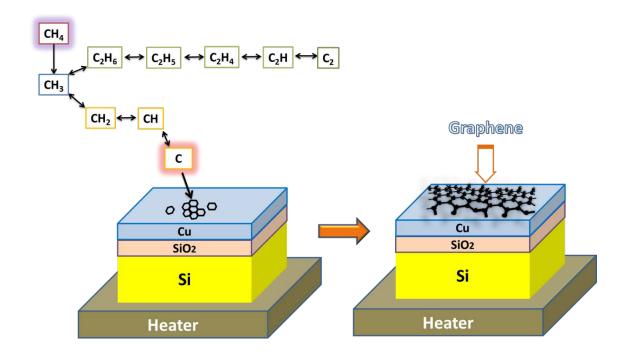
- The thinnest and lightest material known $(1 \text{ m}^2 \sim 0.77 \text{ mg})$
- ➤ The strongest material discovered (100-300 times stronger than steel)
- The best conductor of electricity and heat Extremely high mobility of electrons (more than 2.0 x 10⁵ at room temperature (cm² V⁻¹ s⁻¹)
- Graphene absorbs approximately 2.3% of white light.



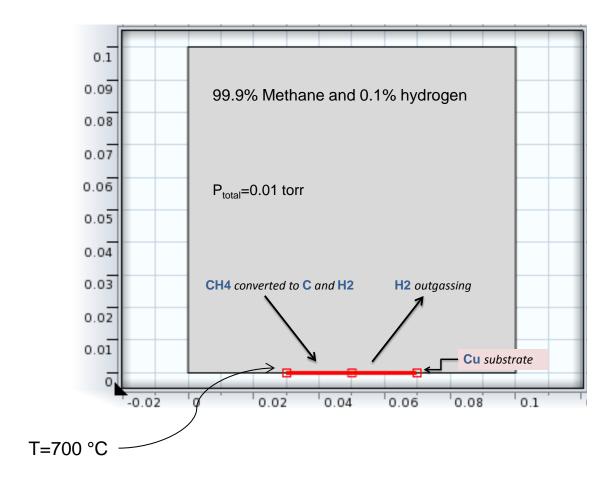


Graphene CVD Growth Mechanism Using Cu

Graphene CVD growth process on Cu thin film



Computational Methods



Computational Methods

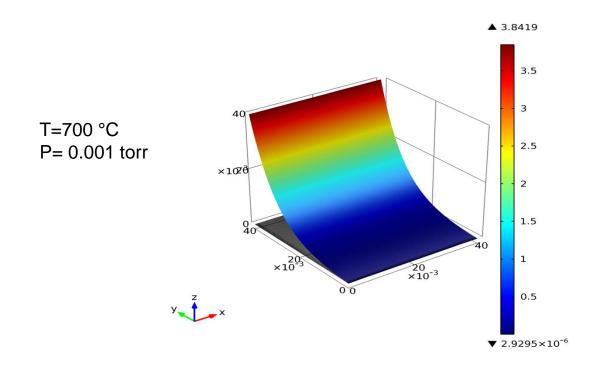
COMSOL model uses Laminar Flow, Heat Transfer, and Heavy Species **Transport** application in order to investigate:

- CH_{Δ} decomposition reactions encountered in the CVD chamber.
- Graphene growth on Cu surface.

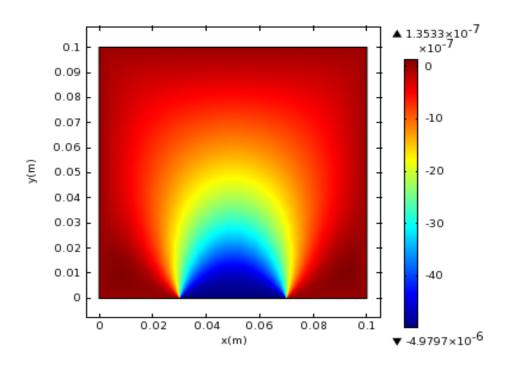
$$\begin{aligned} \mathsf{CH}_4 + \mathsf{5S}_{\,\,(\mathsf{Cu})} &\to \mathsf{graphene}_{\,\,(\mathsf{a})} + 2\mathsf{H}_{2(\mathsf{g})} \\ r_i &= k_i^f \prod_{k=1}^K c_k^{v_{ki}^f} \qquad ; \; \mathsf{The \; surface \; reaction \; rate} \\ \frac{\partial \rho}{\partial t} + \nabla . \left(\rho \boldsymbol{u} \right) = 0 \qquad ; \; \mathsf{The \; continuity \; equation} \\ \rho \frac{\partial \boldsymbol{u}}{\partial t} + \rho \boldsymbol{u}. \, \nabla \boldsymbol{u} &= -\nabla \boldsymbol{p} + \nabla . \left(\mu (\nabla \boldsymbol{u} + (\nabla \boldsymbol{u})^T) - \frac{2}{3} \mu (\nabla . \boldsymbol{u}) \boldsymbol{I} \right) \; ; \; \mathsf{The \; momentum \; balance \; equation} \\ \rho \mathcal{C}_p \frac{\partial T}{\partial t} + \rho \mathcal{C}_p \boldsymbol{u}. \, \nabla T &= \nabla . \left(\kappa \nabla T \right) + Q \qquad \qquad ; \; \mathsf{The \; energy \; balance \; equation} \\ \boldsymbol{u} &= -\frac{M_f}{\rho} \boldsymbol{n} \qquad M_f = \sum_{k=1}^{K_g} M_k \dot{\boldsymbol{s}}_k \qquad \dot{\boldsymbol{s}}_k = \sum_{i=1}^I V_{ki} r_i \end{aligned}$$

The average velocity of the mass species

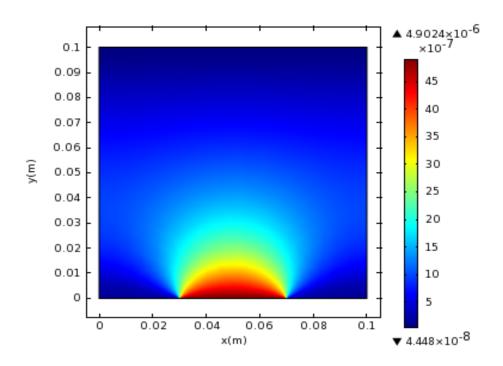
Accumulated graphene growth height (Å) after 1300 sec at H_2 fraction equals 0.001.



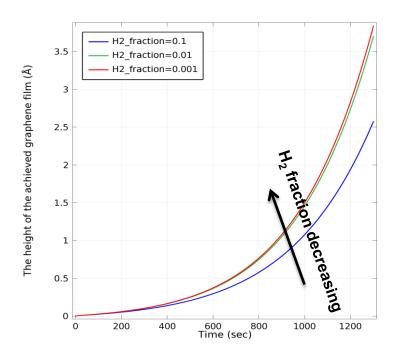
The y- component of the mass averaged velocity at=10 s.



The y- component of the diffusion velocity for hydrogen at=10 s.



The influence of the H₂ fraction upon graphene film thickness.



Conclusion

- □ CVD graphene growth on Cu thin films by direct deposition process has modeled using COMSOL MULTIPHYSICS.
- Laminar Flow, Heat Transfer, and Heavy Species Transport application were utilized to simulate graphene chemical vapor deposition (CVD) growth process on Cu.
- Monolayer graphene film was achieved by methane decomposition at 700
 °C where the total pressure equals ~ 0.01 torr..
- ☐ This model can be extended to simulate graphene growth using different catalysts as well as different carbon precursors.



Thanks

