





TWO-DIMENSIONAL MODEL OF A
LITHIUM IRON-PHOSPHATE SINGLE PARTICLE

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CONTEXT

- Single particle technique employed by Dokko et al.* to investigate the intrinsic electrochemical properties of LiFePO₄
- Spherical secondary single particle of 24 μm composed of a plurality of primary particles with diameter ranging from 100 to 200 nm
- Application of the regular solution model for homogeneous free energy density proposed by Bai et al.** to address phase-separation dynamics under the experimentally relevant conditions of constant applied current

from H. Munakata et al. J. Power Sources 217 (2012) 444

Platinum filament (current collector, ϕ = 20 µm) Glass seal Electrolyte solution Single particle electrode (LiFePO₄)

^{*} K. Dokko et al. J. Power Sources 189 (2009) 783

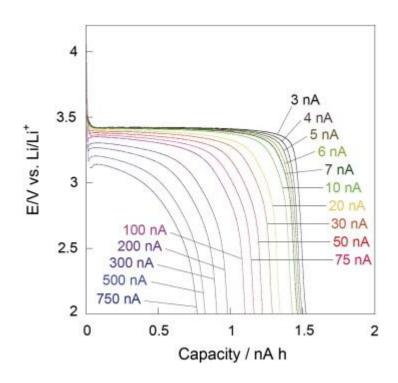
^{**} P. Bai et al. Nano Lett. 11 (2011) 4890





EXPERIMENTAL DATA

- Discharge capacity of the secondary particle at low rate (2C): 1.5 nAh at 30°C
- The particle was discharged galvanostatically at a constant current ranging from 3 nA (2C) to 750 nA (500C) down to 2 V vs. Li/Li+
- Recharge was performed at 3 nA up to 4.2 V
- Even at 500C, more than 50 % of the full capacity was maintained



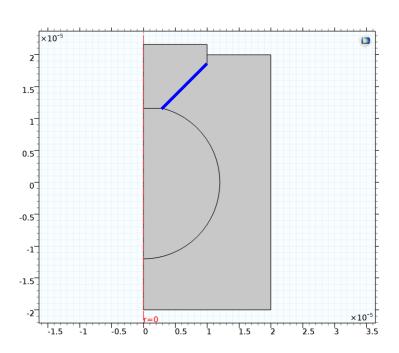
Picture from H. Munakata et al. J. Power Sources 217 (2012) 444





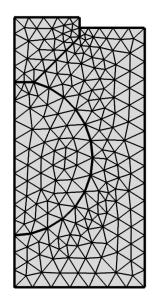
GEOMETRY

2D axisymmetric model

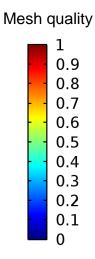


MESH

- Physics-controlled mesh (size: normal)
- Use of a chamfer to improve the mesh











MATERIALS

- Electrolyte (mix 1:1 in volume)
 - Ethylene Carbonate (EC)
 - Propylene Carbonate (PC)
 - Lithium Perchlorate (1M LiClO₄)
- Iron phosphate (FePO₄)
 - Accessible FePO₄ represents only 34 % of the particle volume

Current collector (Pt)



 Electric currents (Ohm's law in solids)

Electric currents
 (Ohm's law in liquids)

 Transport of diluted species (Fick's law for intercalated Li)

 Transport of diluted species (Fick's law for Li⁺ ion)



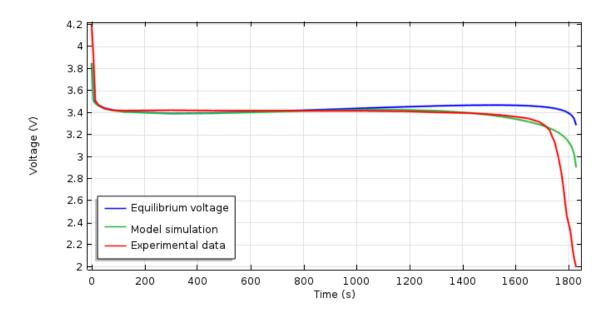








PARTICLE VOLTAGE CURVE IN DISCHARGE AT THE LOWEST RATE (2C)



 The OCV* is derived from the homogeneous chemical potential

$$\mu_{hom} = \Omega \left(1 - 2x_1 \right) + 2k_B T \ln \left(\frac{x_1}{1 - x_1} \right)$$

- with Ω = 0.183 eV in agreement with the LiFePO₄ phase diagram**
- The knee of the particle voltage curve is obtained by considering the apparent volumetric capacity of the LiFePO₄ particle: 34 % of the theoretical volumetric capacity

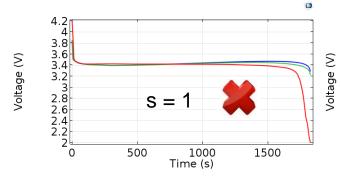
^{*} OCV = Open Circuit Voltage

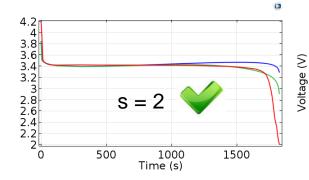
^{**} A. Yamada et al. Electrochem. Solid-State Lett. 8 (2005) A409

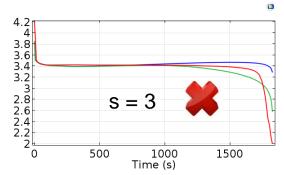


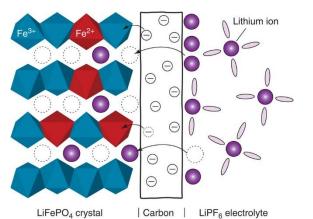


IMPACT OF THE TRANSITION STATE (TS) MODEL ON THE VOLTAGE CURVE AT 2C









 Reaction rate corresponds to a certain choice of model for the activity coefficient of the TS:

$$\mu_{\pm}^{ex} = k_B T \ln \gamma_A = -sk_B T \ln (1 - x_1) + E_{\pm}$$

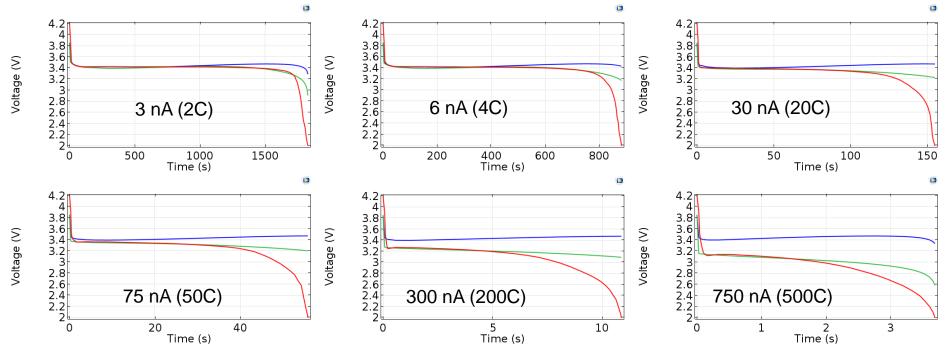
 The transition state may exclude more than one site if we take into account the solvation shell*

> * M. Bazant *Acc. Chem. Res.* **46** (5) (2013) 1144 Picture from Bai et al. *Nat. Commun.* **5** (2014) 3585





PARTICLE VOLTAGE CURVES IN DISCHARGE AT VARIOUS RATES



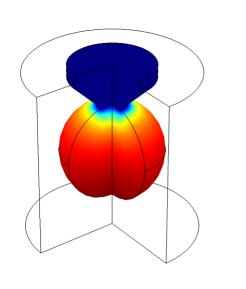
Adjustable parameters: exchange current density, diffusion coefficient in solid phase, material resistance

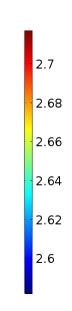




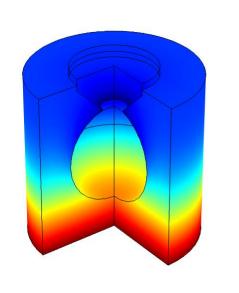
LOCAL POTENTIALS AT THE END OF DISCHARGE AT THE HIGHEST RATE (500C)

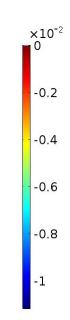
Potential in the solid phases (V)





Potential in the liquid phase



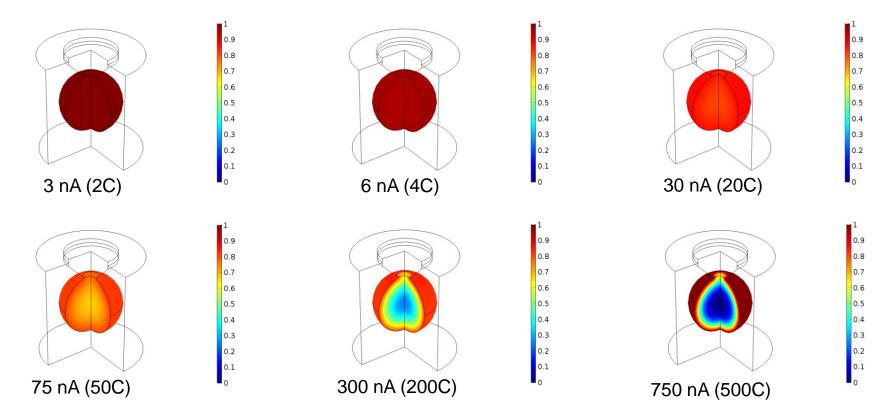


 Conductivity of carbon coated iron phosphate set to 5 mS/cm to fit the experimental voltage Conductivity of electrolyte assumed to be equal to the one of lithium hexafluorophosphate





LOCAL FILLING FRACTION OF THE PARTICLE AT THE END OF DISCHARGE

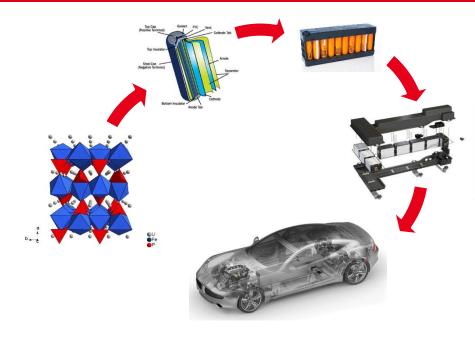






CONCLUSIONS AND PERSPECTIVES

- The 2D axisymmetric model of the LiFePO₄ secondary particle was particularly suitable for evaluating the pertinence of the regular solution model (voltage curve simulations)
- Fast computation time: 15 s for full discharge
- Revolution 2D provides nice cut views of the local particle filling clearly showing a limitation due to Li diffusion in the solid phase at high rate of discharge
- Phase-field model of such a particle could be done successfully with Comsol in a next study involving a better model for both the chemical potential and the transition state



Thank you for your attention!

Do you have any questions?

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